

OPTICALLY PURE (S)- AND (R)-4,5-DIHYDRO-3H-4-METHYLDINAPHTH[2,1-c;
1',2'-e]AZEPINES. APPLICATION TO THE SYNTHESIS OF NEW BIDENTATE
LIGANDS WITH AXIAL ASYMMETRY

Irena G. Stará, Ivo Starý, Jiří Závada*

Institute of Organic Chemistry and Biochemistry
Czechoslovak Academy of Sciences, 166 10 Prague 6, Czechoslovakia

(Received 7 September 1992)

Abstract: Easily available ephedrinium salts (S,1R,2S)- and (R,1R,2S)-2 on treatment with alkoxide base afford enantiomerically pure dihydroazepines (S)- and (R)-1, respectively, in quantitative yields. Cleavage of the corresponding dihydroazepinium quaternary salts (S)- and (R)-4 with N- and S-nucleophiles provides a simple approach to a new series of 1,1'-binaphthalene ligands with two different donor groups in 2,2'-positions.

Axially dissymmetric 1,1'-binaphthyl ligands bearing two identical groups in 2,2'-positions proved to be remarkably useful in enantioselective catalysis, with optical yields close to 100% ee being obtained in several preparatively important reactions¹⁻⁷. Somewhat paradoxically, few axially asymmetric analogues⁸ with non-identical donor groups have been investigated so far, owing possibly to a lack of convenient optically active synthons and/or synthetic procedures.

In this paper, we report a simple synthesis of the optically pure (S)- and (R)-4,5-dihydro-3H-4-methyldinaphth[2,1-c;1',2'-e]azepines, (S)- and (R)-1 (Scheme 1), and of the corresponding quaternary salts, (S)- and (R)-2, which can serve as versatile building blocks for new optically active 1,1'-binaphthalenes with donor groups in 2,2'-positions.

It is known that the easily accessible racemic 2,2'-bis(bromomethyl)-1,1'-binaphthalene 4 reacts smoothly with *l*-ephedrine yielding a diastereoisomeric mixture of the quaternary salts (S,1R,2S)- and (R,1R,2S)-5 which is readily separated by crystallization⁹ (Scheme 2). As we have now found, treatment of the individual ephedrinium salts with alkoxide base leads quantitatively to the optically pure dihydroazepines (S)- and (R)-1 (Scheme 2)^{10,11}, which on alkylation afford the corresponding quaternary salts (S)- and (R)-2, respectively, again in very high yields (Table 1).

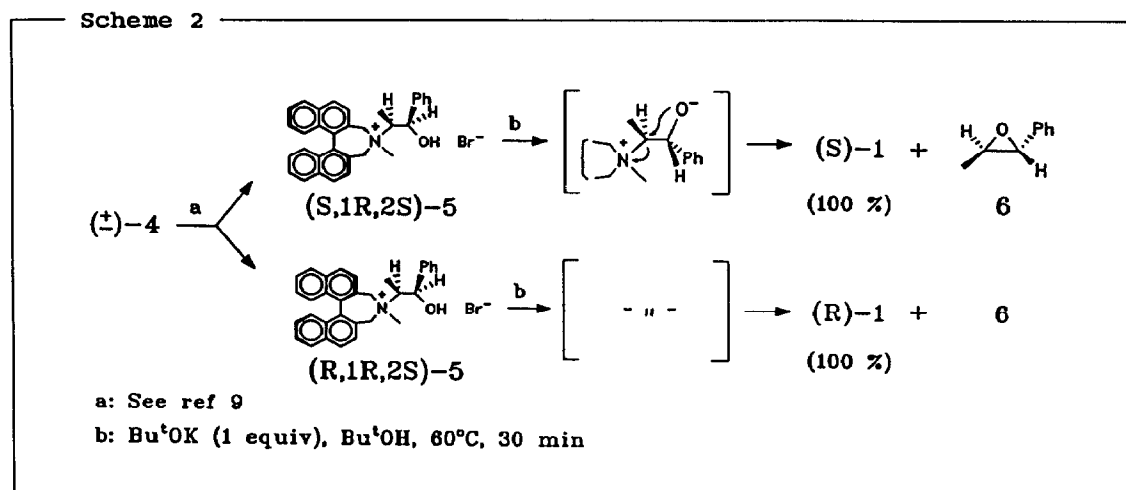
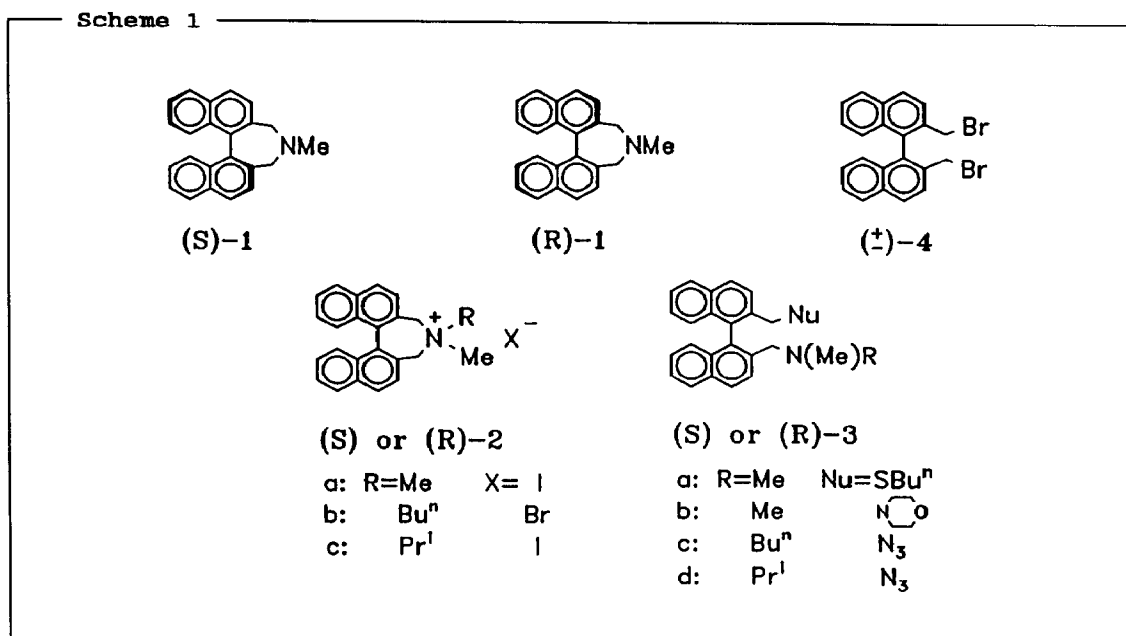


Table 1. Preparation of Optically Active Quaternary Salts 2

educt	conditions ^a	product ^b	yield ^c (%)	[α] _D ²² (c) ^d	mp (°C)
(S)-1	A	(S)-2a	99	+314 (0.42)	197-9
(R)-1	A	(R)-2a	99	-315 (0.41)	197-200
(S)-1	B	(S)-2b	94	+268 (0.51)	178-81
(S)-1	C	(S)-2c	85	+261 (0.94)	253-6 dec

^aA, MeI (1.0 equiv), benzene, rt, overnight; B, BuⁿBr (2.0 equiv), benzene, reflux, 6 h; C, Pr^lI (5.0 equiv), benzene, reflux, 15 h.
^bSee ref 18 for physical data. ^cIsolated. ^dIn DMSO.

Retrosynthetic analysis suggests several alternatives for transformation of the compound 4 into 2,2'-bifunctional 1,1'-binaphthalenes. As we already pointed out elsewhere¹², racemic quaternary salts 2 react with a variety of nucleophiles (e.g. amines, azide, malonate, mercaptide, phosphide, and selenide) preferentially at the benzylic carbon affording the corresponding 2,2'-disubstituted 1,1'-binaphthalenes 3 in good chemical yields. Viability of this procedure for the preparation of the optically active¹¹ derivatives (S)- and (R)-3 has now been examined and some successful examples are recorded in Table 2.

Table 2. Cleavage of Quaternary Salts 2 with Nucleophiles

educt	conditions ^a	product ^b	yield ^c (%)	$[\alpha]_D^{22}$ (c)
(R)-2a	A	(R)-3a	78	+117 (0.42) ^d
(R)-2a	B	(R)-3b	83	+ 69 (0.27) ^e
(S)-2b	C	(S)-3c	64	- 87 (0.55) ^e
(S)-2c	C	(S)-3d	94	- 94 (0.28) ^e

^aA, BuⁿSH (2.2 equiv), Bu^tOK (2.2 equiv), EtOH, 65°C, 2 h; B, morpholine (excess), 120°C, 3 h; C, NaN₃ (1.5 equiv), DMF, 120°C, 1 h. ^bSee ref 18 for physical data. ^cIsolated. ^dIn benzene. ^eIn CHCl₃.

In this way, a straightforward access to a new series of optically active 1,1'-binaphthalene bidentate ligands with two different donor groups has been provided.

Acknowledgement: A financial support from the Grant Agency of Czechoslovak Academy of Sciences (Reg. 45505) is gratefully acknowledged. We thank Dr. P. Fiedler for obtaining and interpreting IR spectra. In addition, we are grateful to Dr. D. Šaman for checking optical purity of selected compounds by ¹H NMR.

References and Notes

- Noyori, R.; Takaya, H. *Acc. Chem. Res.* 1990, 23, 345.
- Noyori, R.; Suga, S.; Kawai, K.; Okada, S.; Kitamura, M. *Pure Appl. Chem.* 1988, 60, 1597.
- Tomioka, K. *Synthesis* 1990, 541.
- Brunner, H. *Synthesis* 1988, 645.
- Narasaka, K. *Synthesis* 1991, 1.
- Noyori, R. *Chem. Soc. Rev.* 1989, 18, 187.
- Kaufmann, D.; Boese, R. *Angew. Chem. Int. Ed. Eng.* 1990, 29, 545.
- According to a customary stereochemical analysis, axially dissymmetric bidentate ligands (C₂ symmetry) are expected to be more effective in reducing the number of possible diastereoisomeric transition states than the corresponding axially asymmetric analogues (C₁ symmetry)¹³. In practice, however, bidentate ligands with identical donor groups are often less effective than those bearing two different donor groups^{3,4,14-16}.
- Maigrot, N.; Mazaleyrat, J.P. *Synthesis* 1985, 317.
- The accompanying epoxide 6 has been identified by a comparison with an authentic sample, cf. ref 17.
- Optical purity of compounds (S)-1, (R)-1, (R)-3a, and (R)-3b has been confirmed by NMR using (S)-(+)-2,2,2-trifluoro-1-(9-anthryl)-ethanol in CDCl₃. The absence of configurational scrambling in azide anion cleavage of dihydroazepinium salts has been proved earlier¹².

12. Stará, I.G.; Stary, I.; Závada, J. *J. Org. Chem.* 1992, in press.
13. Whitesell, J.K. *Chem. Rev.* 1989, 89, 1581.
14. Hayashi, T.; Konishi, M.; Fukushima, M.; Kanekira, K.; Hioki, T.; Kumada, M. *J. Org. Chem.* 1983, 48, 2195.
15. Oishi, T.; Hiram, M. *Tetrahedron Lett.* 1992, 33, 639.
16. Corey, E. J. *Pure Appl. Chem.* 1990, 62, 1209.
17. Witkop, B.; Foltz, C.M. *J. Am. Chem. Soc.* 1957, 79, 197.
18. Physical Data for New Compounds:

(S)-1: mp 151-3°C, $[\alpha]_D^{22} +446$ (c 0.47, DMSO). $^1\text{H NMR}$ (200 MHz, DMSO- d_6) δ 2.26 (s, 3 H, CH_3), 2.94 and 3.60 (AB q, $J=12.1$ Hz, 4 H, ArCH_2N^+), 7.23-8.10 (m, 12 H, arom). IR (KBr) 2792 $\nu_{\text{S}}(\text{CH}_3)$ cm^{-1} . EI MS m/z (rel. inten.) 309 (M^+ , 37), 294 (6), 266 (100), 252 (7).

(R)-1: mp 151.5-152.5°C, $[\alpha]_D^{22} -445$ (c 0.46, DMSO).

(S)-2a: $^1\text{H NMR}$ (200 MHz, DMSO- d_6) δ 3.13 (s, 6 H, 2x CH_3), 3.92 and 4.57 (AB q, $J=13.2$ Hz, 4 H, 2x ArCH_2N^+), 7.29-8.32 (m, 12 H, arom). IR (KBr) 2999 $\nu_{\text{S}}(\text{CH}_3)$, 1473 and 1466 $\delta_{\text{AS}}(\text{CH}_3)$ cm^{-1} . FAB MS (thioglycerol/glycerol (3:1) matrix) m/z (rel. inten.) 324 (M^+-I , 100), 281 (29), 266 (12), 252 (4).

(S)-2b: $^1\text{H NMR}$ (200 MHz, DMSO- d_6) δ 0.94 (t, $J=7.0$ Hz, 3 H, CH_2-CH_3), 1.22-1.43 (m, 2 H, CH_2-CH_3), 1.63-2.03 (m, 2 H, $\text{CH}_2-\text{CH}_2-\text{CH}_3$), 3.08 (s, 3 H, N^+-CH_3), 3.15-3.51 (m, overlapped by signal of water from DMSO- d_6 , 2 H, $\text{N}^+-\text{CH}_2-\text{CH}_2$), 3.87 and 4.63 (AB q, $J=13.1$ Hz, 2 H, ArCH_2N^+), 3.91 and 4.63 (AB q, $J=13.1$ Hz, 2 H, ArCH_2N^+), 7.22-8.35 (m, 12 H, arom). IR (KBr) 3003 $\nu_{\text{AS}}(\text{CH}_3)$ of CH_3N^+ , 1370 $\delta_{\text{S}}(\text{CH}_3)$ of Bu^{n} cm^{-1} . FAB MS (thioglycerol/glycerol (3:1) matrix) m/z (rel. inten.) 366 (M^+-Br , 100), 308 (2), 281 (15), 266 (11), 252 (2).

(S)-2c: $^1\text{H NMR}$ (200 MHz, DMSO- d_6) δ 1.35 (d, $J=6.1$ Hz, 3 H, $\text{CH}-\text{CH}_3$), 1.62 (d, $J=6.1$ Hz, 3 H, $\text{CH}-\text{CH}_3$), 2.88 (s, 3 H, N^+-CH_3), 3.49-3.67 (m, 1 H, $\text{CH}_3-\text{CH}-\text{CH}_3$), 3.71 and 4.78 (AB q, $J=13.5$ Hz, 2 H, ArCH_2N^+), 3.90 and 4.63 (AB q, $J=12.5$ Hz, 2 H, ArCH_2N^+), 7.20-8.40 (m, 12 H, arom). IR (KBr) 3005 $\nu_{\text{AS}}(\text{CH}_3)$ of CH_3N^+ , 1387 and 1369 $\delta_{\text{S}}(\text{CH}_3)$ of Pr^{I} cm^{-1} . FAB MS (thioglycerol/glycerol (3:1) matrix) m/z (rel. inten.) 352 (M^+-I , 100), 308 (3), 281 (26), 266 (14), 252 (3).

(R)-3a: $^1\text{H NMR}$ (200 MHz, CDCl_3/TMS) δ 1.01 (s, 7 H, $-\text{CH}_2-\text{CH}_2-\text{CH}_3$), 1.31 (s, 2 H, SCH_2-CH_2), 2.08 (s, 6 H, 2x CH_3), 2.93 and 3.46 (AB q, $J=14.5$ Hz, 2 H, ArCH_2N), 3.41 (s, 2 H, ArCH_2S), 7.00-8.00 (m, 12 H, arom). IR (CCl_4) 2819 and 2772 $\nu_{\text{S}}(\text{CH}_3)$ of $(\text{CH}_3)_2\text{N}$ cm^{-1} . FAB MS (thioglycerol/glycerol (3:1) matrix) m/z (rel. inten.) 414 (M^++1 , 100), 356 (10), 313 (45), 279 (97), 267 (75), 252 (20).

(R)-3b: $^1\text{H NMR}$ (200 MHz, CDCl_3/TMS) δ 2.06 (s, 6 H, 2x CH_3), 2.19 (m, $W/2=11.5$ Hz, 4 H, $\text{NCH}_2\text{CH}_2\text{O}$), 2.93 and 3.05 (AB q, $J=14.2$ Hz, 2 H, CH_2 -benzyl), 3.23 and 3.30 (AB q, $J=14.4$ Hz, 2 H, CH_2 -benzyl), 3.56 (t, $J=4.5$ Hz, 4 H, $\text{NCH}_2\text{CH}_2\text{O}$), 6.91-8.02 (m, 12 H, arom). IR (CCl_4) 2818 and 2771 $\nu_{\text{S}}(\text{CH}_3)$, 1119 $\nu_{\text{AS}}(\text{C}-\text{O}-\text{C})$ cm^{-1} . EI MS m/z (rel. inten.) 410 (M^+ , 20), 365 (22), 323 (25), 279 (100), 265 (25), 252 (5).

(S)-3c: $^1\text{H NMR}$ (200 MHz, CDCl_3) δ 0.81 (t, $J=6.8$ Hz, 3 H, CH_2-CH_3), 1.07-1.34 (m, 4 H, $\text{CH}_2-\text{CH}_2-\text{CH}_3$), 1.99-2.28 (m, 2 H, $\text{N}-\text{CH}_2-\text{CH}_2$), 2.00 (s, 3 H, $\text{N}-\text{CH}_3$), 3.00 and 3.25 (AB q, $J=14.0$ Hz, 2 H, ArCH_2N), 4.08 (s, 2 H, ArCH_2N_3), 6.95-8.08 (m, 12 H, arom). IR (CCl_4) 2791 $\nu_{\text{S}}(\text{CH}_3)$ of CH_3N , 2099 $\nu_{\text{AS}}(\text{N}_3)$, 1378 $\delta_{\text{S}}(\text{CH}_3)$ of Bu^{n} cm^{-1} . EI MS m/z (rel. inten.) 407 (M^+-1 , 1.6), 393 (1.1), 380 (24), 365 (13), 352 (7), 323 (5), 294 (100), 280 (99), 266 (95), 252 (23).

(S)-3d: $^1\text{H NMR}$ (200 MHz, CDCl_3/TMS) δ 0.70 (d, $J=6.6$ Hz, 3 H, $\text{CH}-\text{CH}_3$), 0.75 (d, $J=6.6$ Hz, 3 H, $\text{CH}-\text{CH}_3$), 1.99 (s, 3 H, $\text{N}-\text{CH}_3$), 2.49-2.71 (m, 1 H, $\text{CH}-\text{CH}_3$), 3.09 and 3.30 (AB q, $J=14.0$ Hz, 2 H, ArCH_2N), 4.09 (d, $J=1.4$ Hz, 2 H, ArCH_2N_3), 6.98-8.13 (m, 12 H, arom). IR (CCl_4) 2789 $\nu_{\text{S}}(\text{CH}_3)$ of CH_3N , 2099 $\nu_{\text{AS}}(\text{N}_3)$, 1381 and 1362 $\delta_{\text{S}}(\text{CH}_3)$ of Pr^{I} cm^{-1} . FAB MS (glycerol/TFA matrix) m/z (rel. inten.) 395 (M^++1 , 100), 352 (4), 351 (4), 323 (7), 308 (7), 294 (53), 280 (26), 266 (53), 252 (41).